METAALICUS PROJECT

(Mercury Experiment To Assess Atmospheric Loading in Canada and the US)

Relating Mercury Deposition to Methylmercury Levels in Fish

Presentation by

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EPRI

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Public and Private Sector Participation

- Fisheries and Oceans Canada
- USGS
- EPRI
- University of Wisconsin/WDNR
- US EPA
- US DOE
- Academy of Natural Sciences, MD
- Minnesota Pollution Control Agency
- U. Toronto, U. Quebec, Trent U., U. Alberta
- NSERC
- Environment Canada
- Tetra Tech Inc.
- Swedish IVL institute

Principal Investigators:

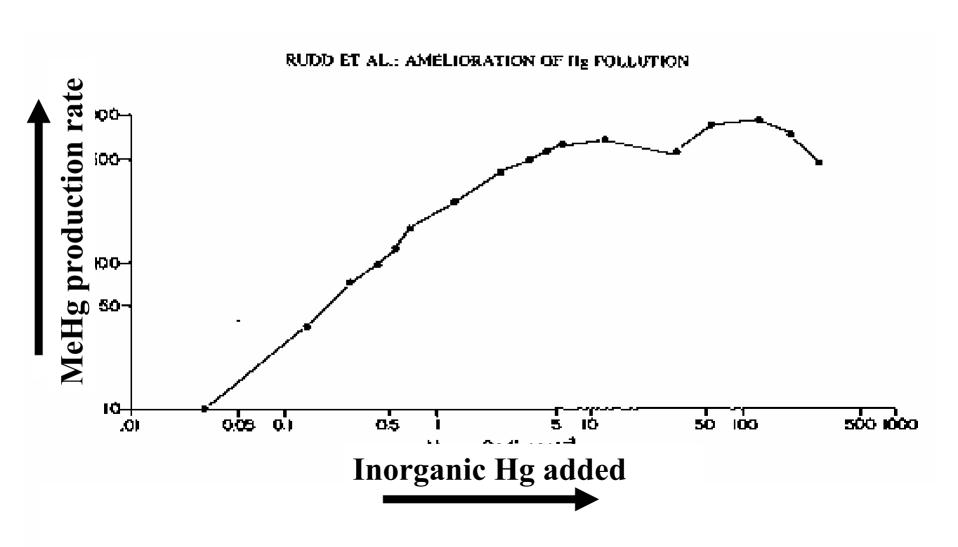
The Freshwater Institute: Ken Beaty, Paul Blanchfield, Drew Bodaly, Carol Kelly, Mike Paterson, Cheryl Podemski, John Rudd, Michael Turner The Academy of Natural Sciences: Cindy Gilmour Tetra Tech: Reed Harris Trent Univ.: Holger Hintelmann **USGS**: Dave Krabbenhoft US DOE: Steve Lindberg U. Alberta: Vince St. Louis U. Maryland: Rob Mason, Andrew Heyes U. Montreal: Marc Amyot U. Toronto: Brian Branfireun U. Wisconsin: Jim Hurley, Chris Babiarz

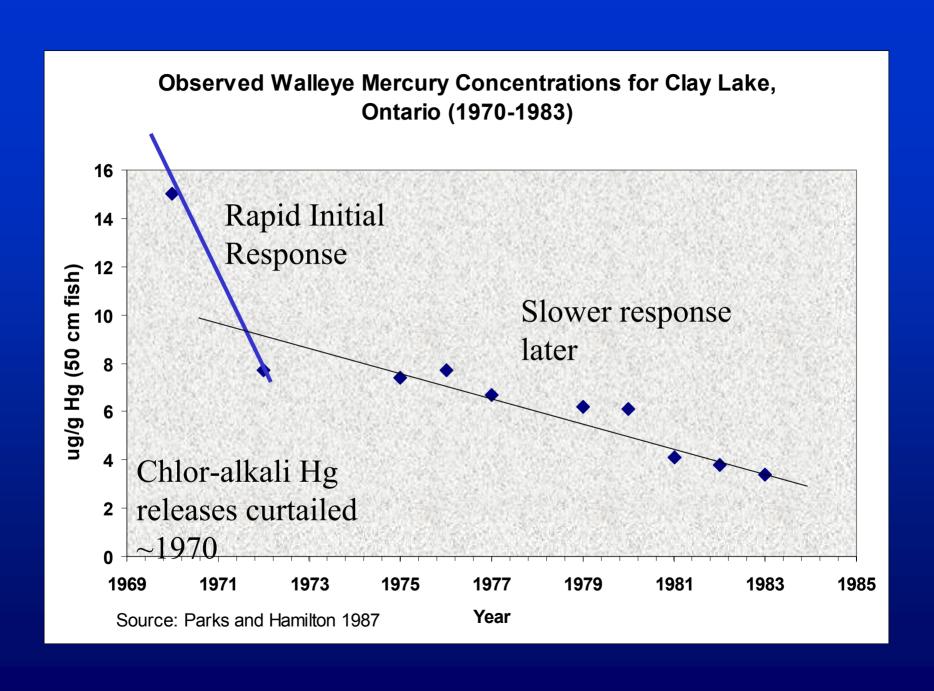
Fundamental Questions to be addressed by



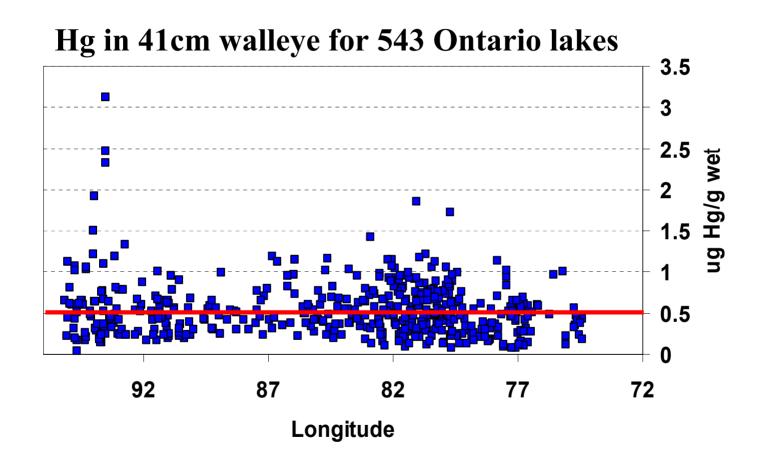
- What is the relationship between the amount of Hg in atmospheric deposition and the amount of methylmercury in fish?
- How quickly will the fish Hg levels respond to a change/reduction in mercury deposition?
- How will environmental factors affect the magnitude and timing of the response?

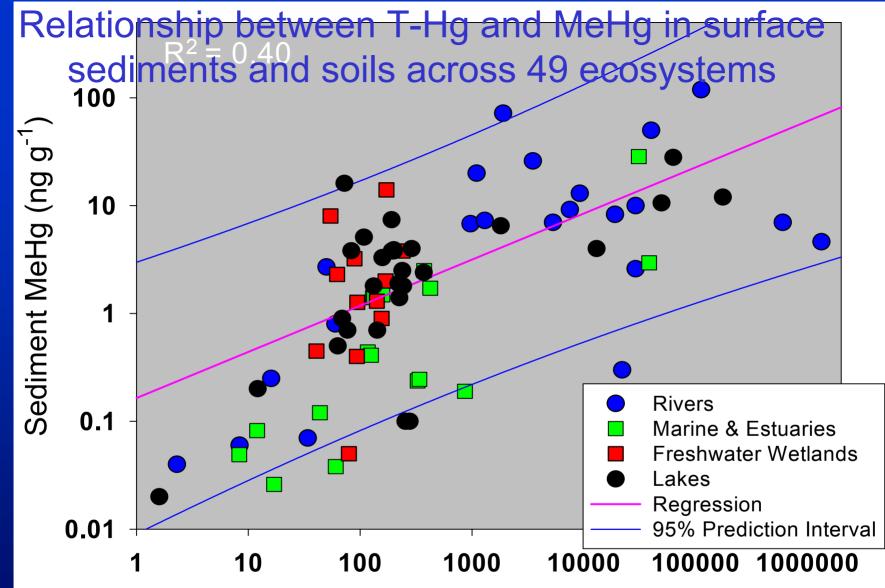
We have limited laboratory and mesocosm data, but what would happen in the real world?





Many of these lakes have similar atmospheric Hg deposition... Does that mean loading is not a factor?





Benoit, J., C. Gilmour, A. Heyes, R.P. Mason, C. Miller. 2003. Geochemical and Biological Controls Over Methylmercus Ediment T-Hg (ng g⁻¹)

Production and Degradation in Aquatic Ecosystems. In:

ACS Symposium Series #835, Y. Chai and O.C. Braids, Eds

American Chemical Society, Washington, DC. pp. 262-297

[&]quot;Biogeochemistry of Environmentally Important Trace Elements",

Methylmercury production varies widely among ecosystems

- There is a significant relationship between Hg and MeHg in sediments across ecosystems implying that reduction in Hg deposition will lessen MeHg production.
- However, other factors contribute substantially to the variability in MeHg production within and among ecosystems.
- Mercury load alone cannot be used to predict MeHg production with much confidence.

Where does Hg methylation occur?

In wetlands:

- Surface sediments/soils
- Periphyton

In lakes:

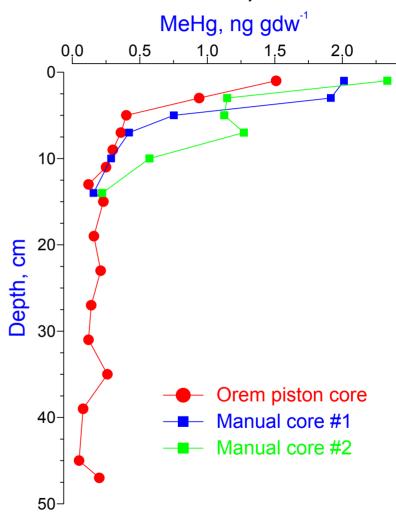
- Surface sediments
- Oxycline

In watersheds:

Saturated soils

Sewage Treatment Plants

Everglades WCA 2B Site 2BS, 12/96



Ecosystems prone to high MeHg:

- Freshwater wetlands
- Reservoirs
- Freshwater ecosystems affected by acid deposition
- Ecosystems affected by Hg

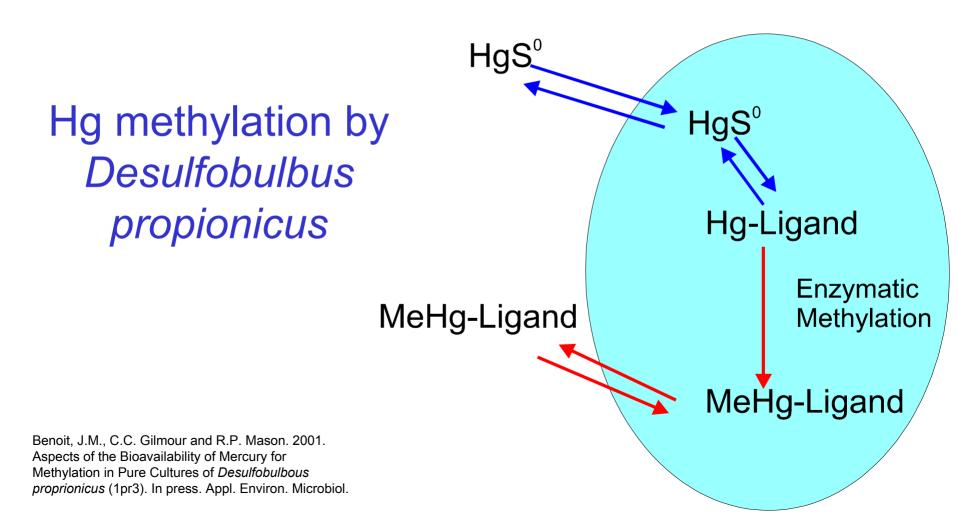
Variability in MeHg production among ecosystems is a function of:

- Hg loading
 - Atmospheric
 - Point sources
- Basin geomorphology
 - Wetland area
 - Littoral area
 - Watershed area

- Sulfur loading
 - Acid deposition
 - Agriculture
 - Natural sources

- Controls on microbial activity
- Temperature
- Trophic status

- Hg methylation is mediated by sulfate-reducing bacteria
- Methylation occurs inside cells
- Inorganic Hg speciation determines uptake rates by cells



What is METAALICUS?

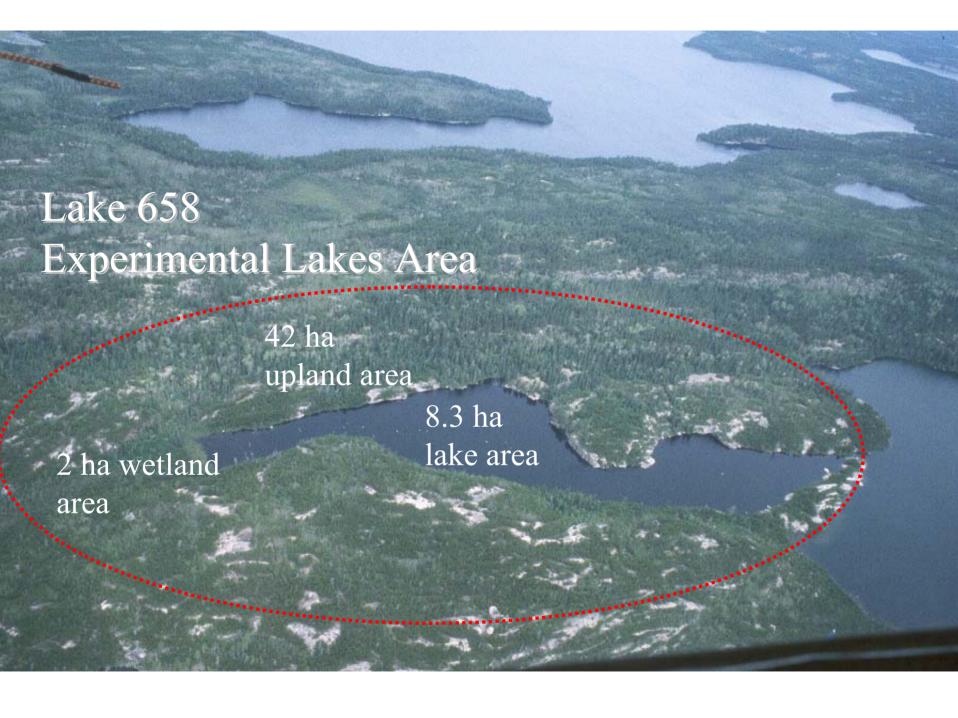
- A loading experiment: Mercury is being added to a lake and its surrounding watershed.
- Wet Hg deposition rate is being increased by 4X
- Stable Hg(II) isotopes (non-radioactive) are being used



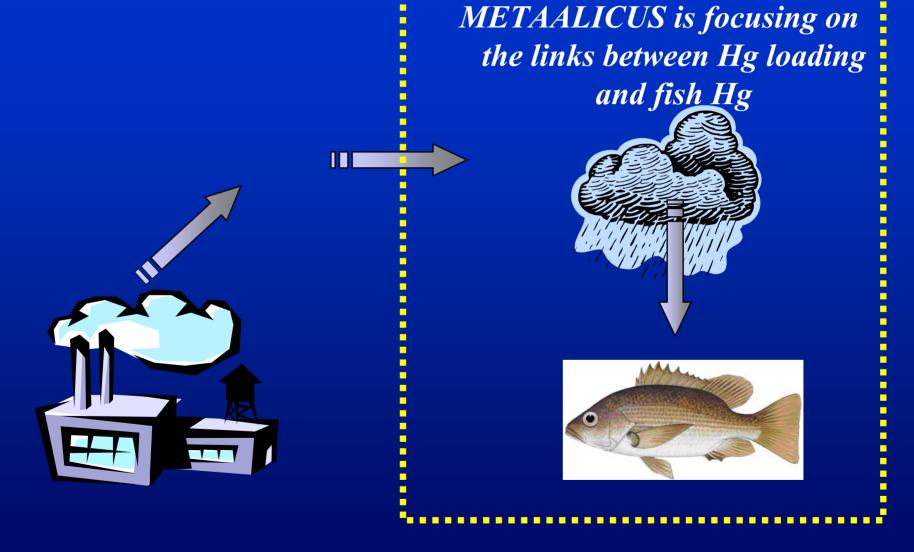


The Experimental Lakes Area Research Facility

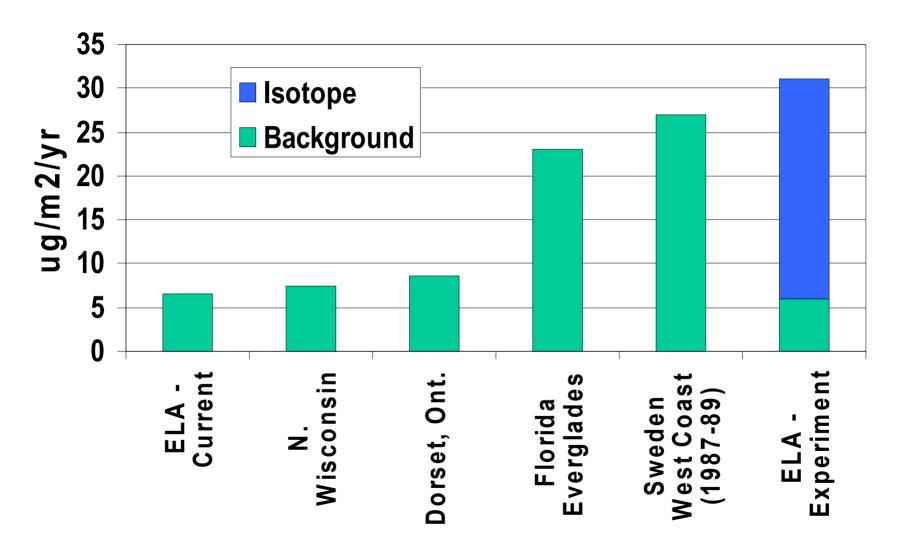


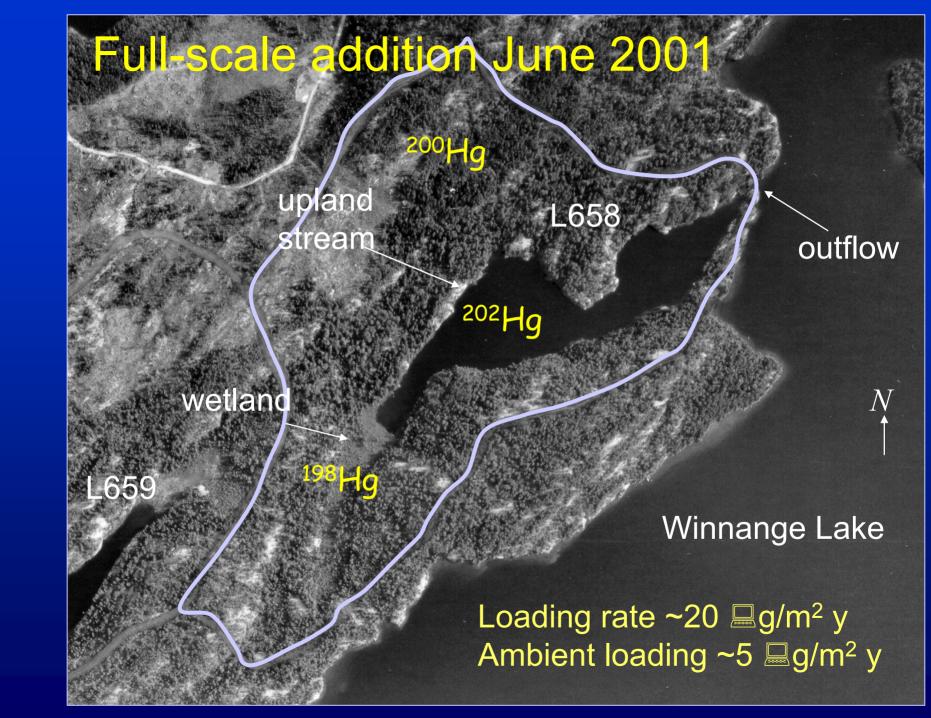


METAALICUS Scope

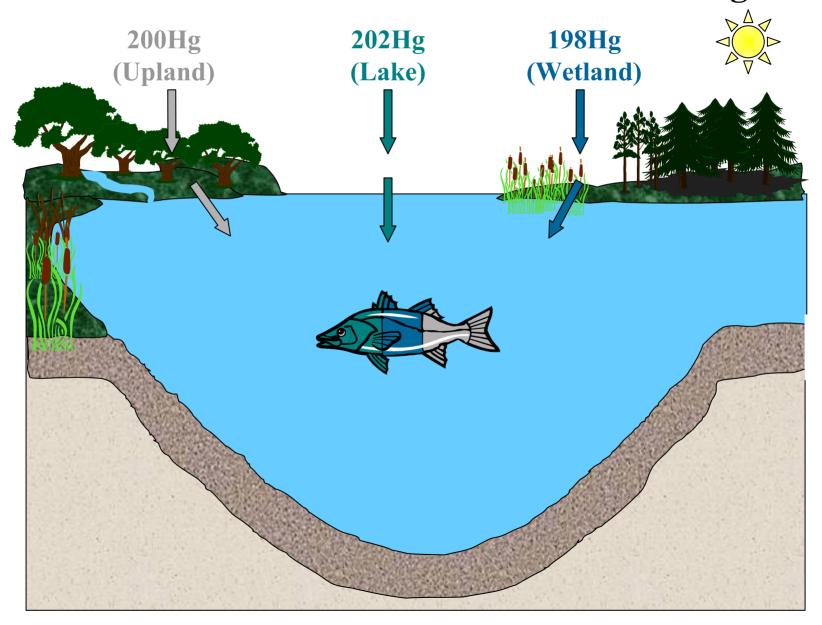


How much will wet Hg deposition be increased?





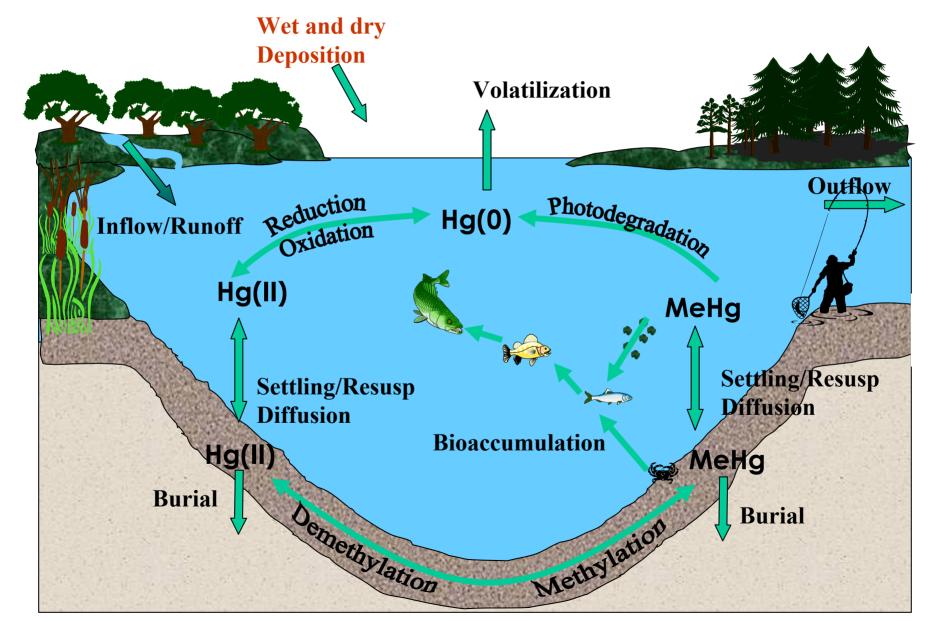
Measure Different Contributions to Fish Hg



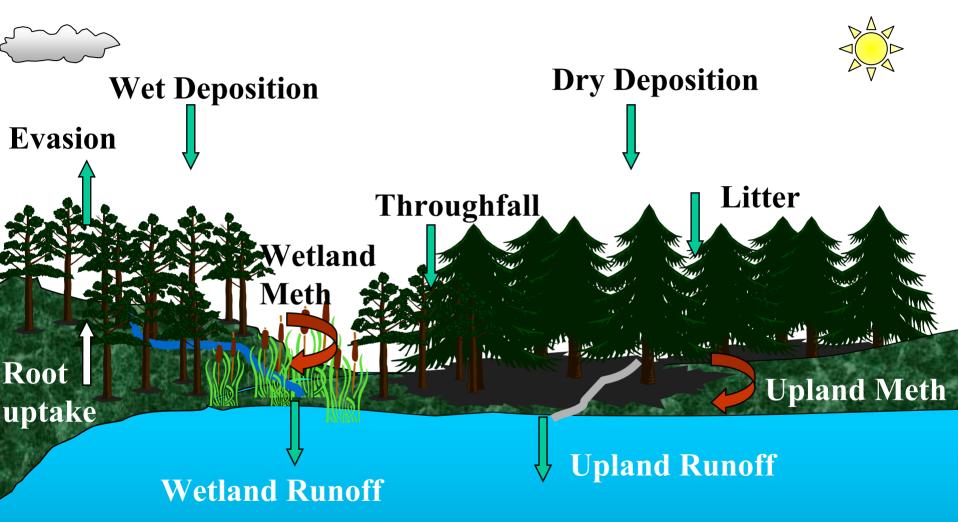


METAALICUS is looking at





METAALICUS in the watershed



Pilot studies: How much of the added Hg is exported in runoff from an upland catchment?

Mist application to represent light rain or dry deposition:

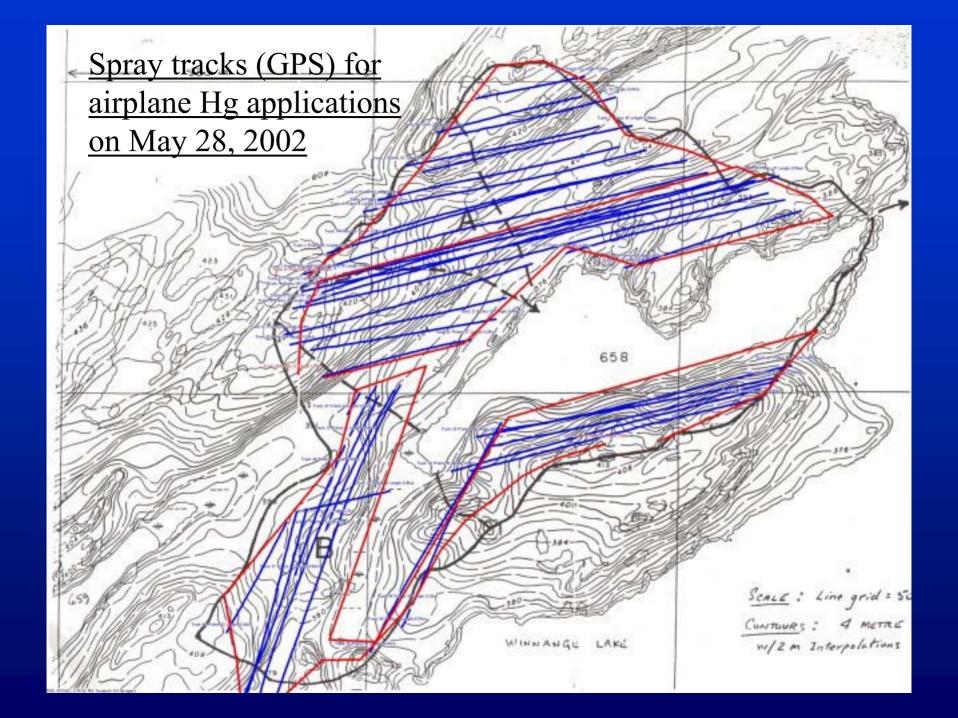


- 0.13% in 1999 (one spike ²⁰²Hg)
- 0.06% in 2000 (multiple ²⁰⁰Hg spikes)

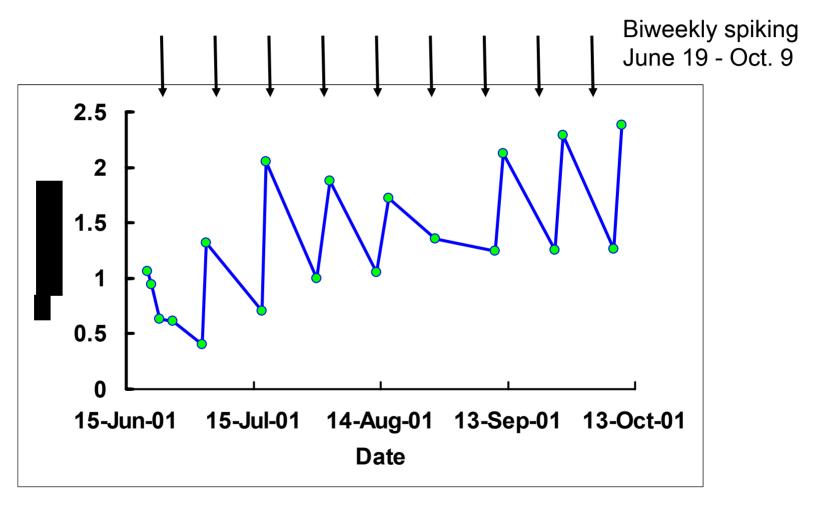
Simulated rain application to create runoff flow:



~10-17% (Hg, MeHg) of spike in runoff.

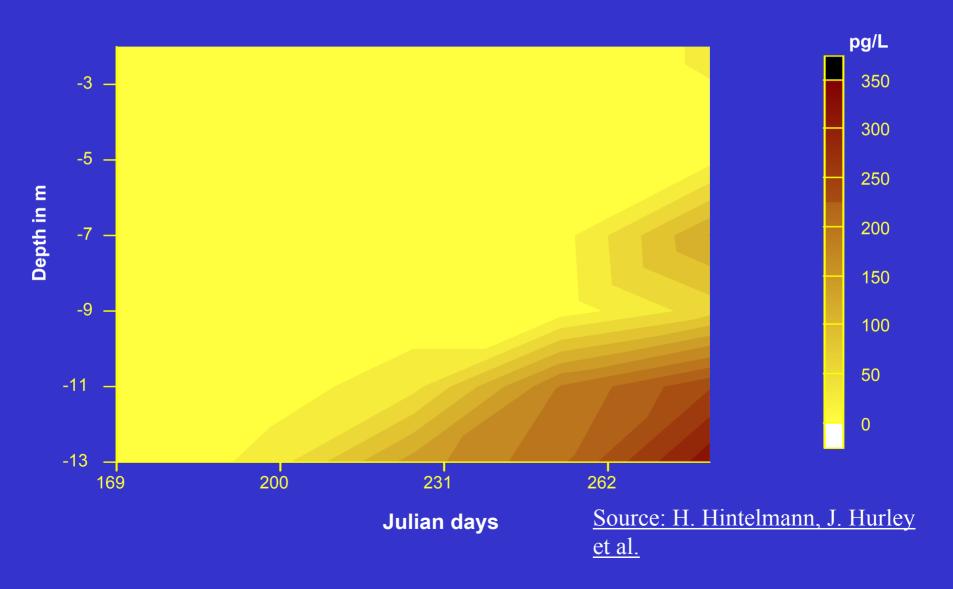


Lake 658 spikes with ²⁰²Hg in 2001



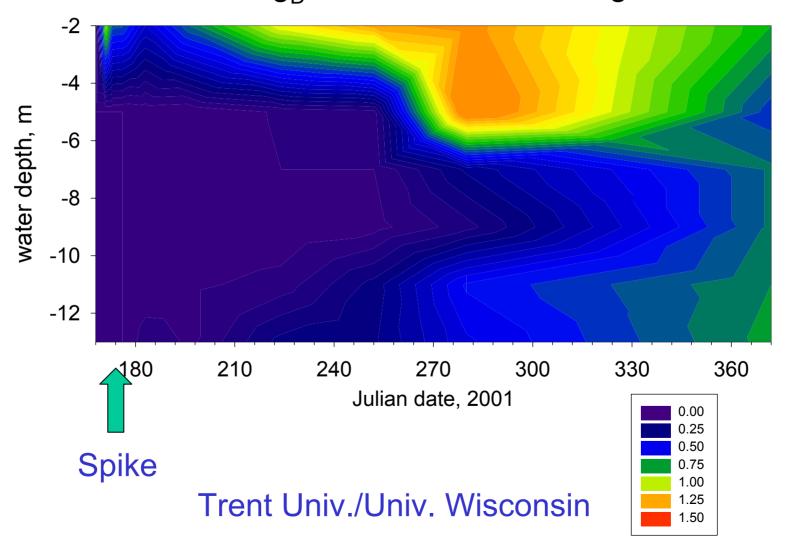
Source: H. Hintelmann

202-MeHg dissolved in Lake 658 June 18 - October 8, 2001



²⁰²Hg in lake water

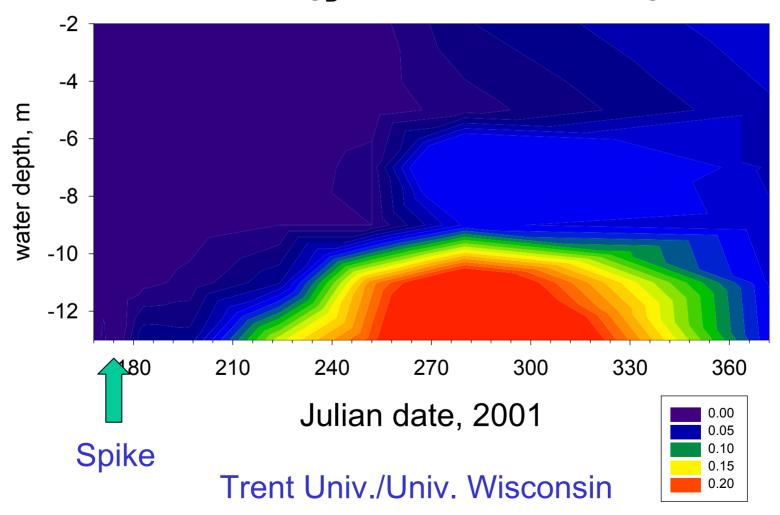
By September, about 30% of Hg in lake water is ²⁰²Hg_D in L658 lake water, ng/L

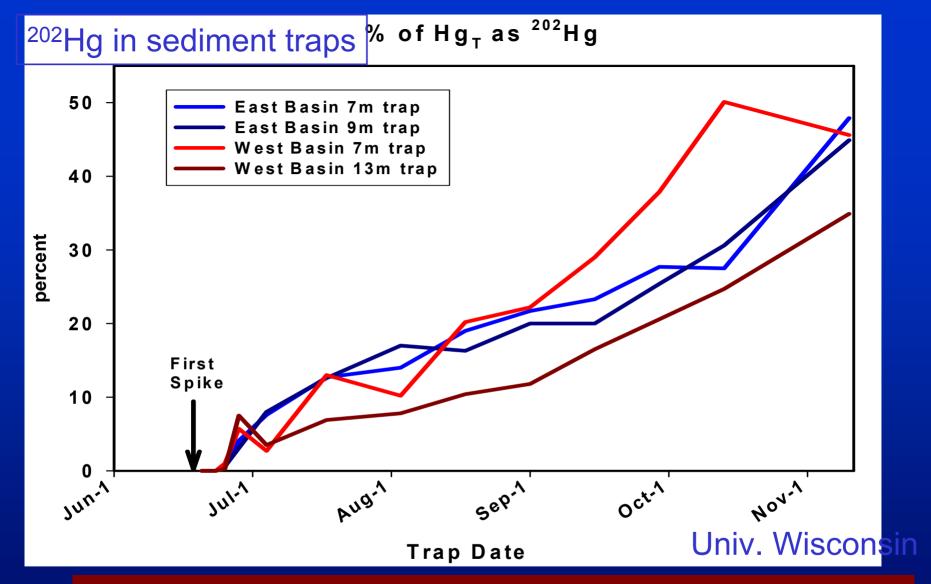


Me²⁰²Hg in lake water

By September, about 15% of MeHg in lake water is Me²⁰²Hg

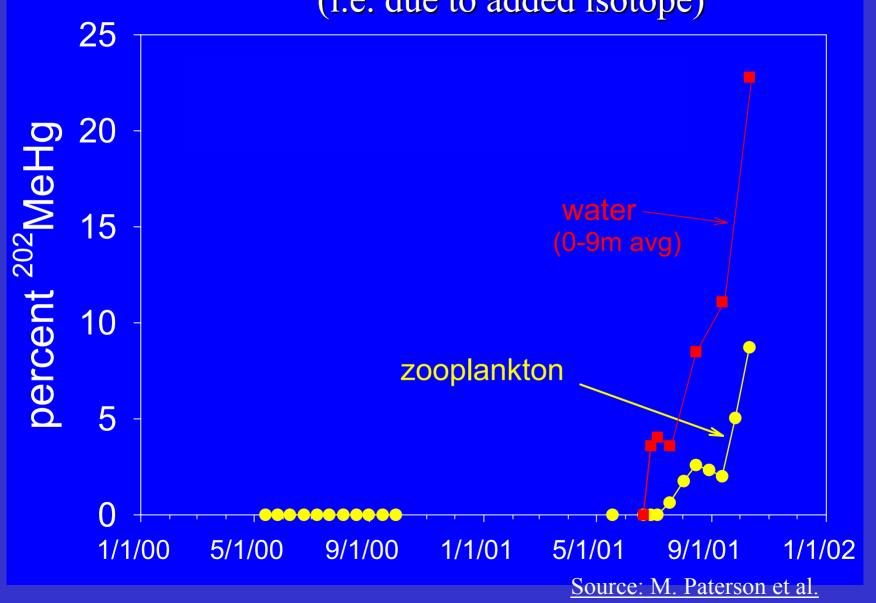
Me²⁰²Hg_D in L658 lake water, ng/L



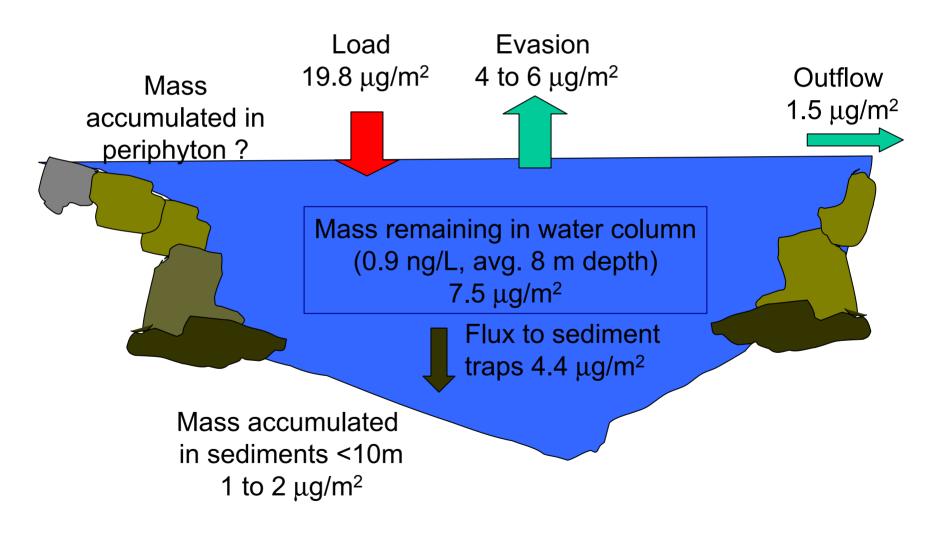


Within <u>6 days</u> of spiking, ²⁰²Hg began appearing in particles in sediment traps and by late summer, comprised <u>almost</u> <u>50%</u> of all Hg in sinking particles.

Percent of measured MeHg as excess 202MeHg (i.e. due to added isotope)

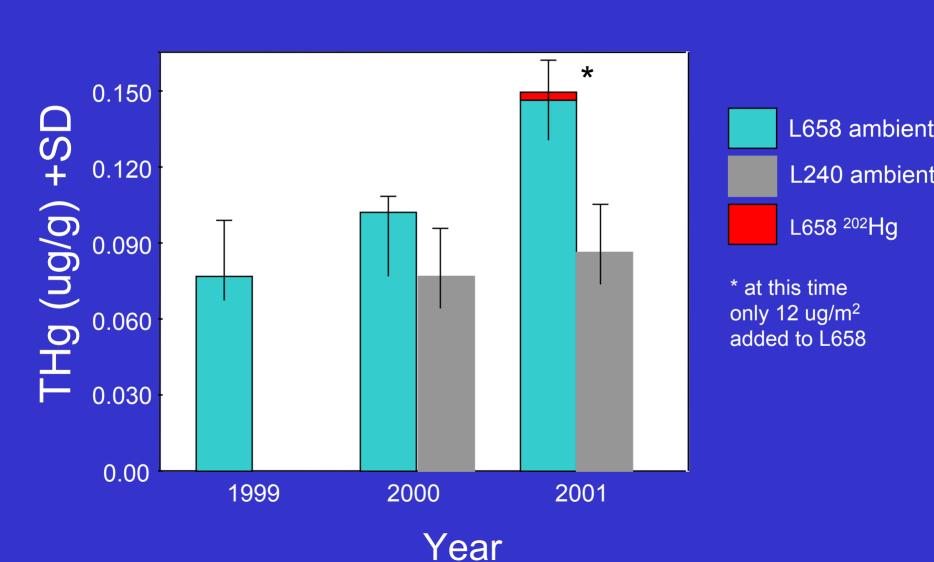


Preliminary ²⁰²Hg Budget for Lake 658 (June – October 2001)



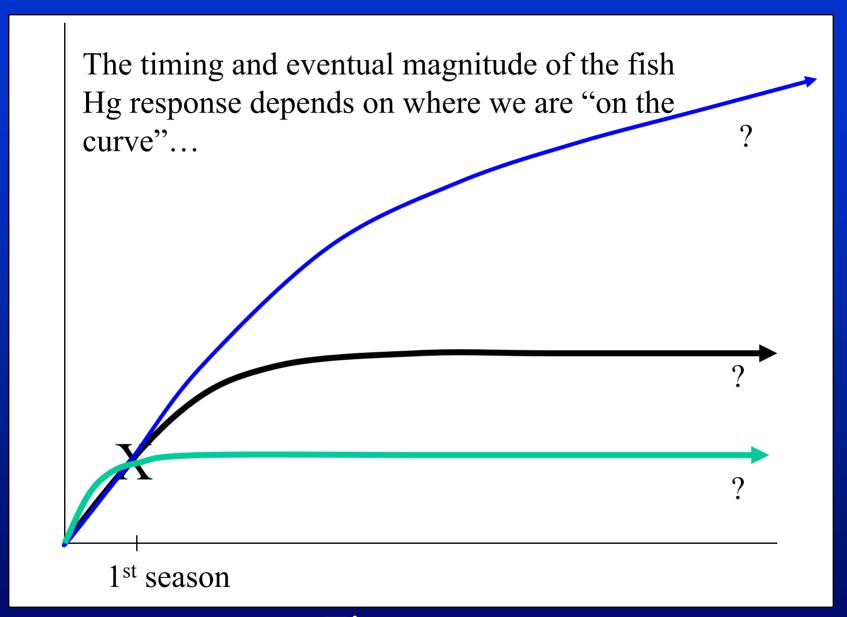
Hg in young of year perch





What does it all mean so far?...

- 1. The conversion from Hg(II) to MeHg of isotope added directly to the lake began quickly (within weeks)
- 2. Too soon in the experiment to say where the isotopic levels will end up in fish, and when...



Time